

## Questions – Chapter 01

1- Who is considered as the “father” of mass spectrometry?

- ☐ Thomson                      ☐ Tanaka                      ☐ Watson                      ☐ Makarov

2- What amino acid(s) could be phosphorylated?

- ☐ Glycine                      ☐ Serine                      ☐ Threonine                      ☐ Tryptophan

3- What is typical of protein secondary structure?

- ☐ Disulfide bridges                      ☐  $\beta$ -sheet                      ☐  $\alpha$ -helix                      ☐ Hydrophobic interactions

4- What amino acid(s) present a charged side chain?

- ☐ Arginine                      ☐ Lysine                      ☐ Glycine                      ☐ Leucine

5- What is single-letter code for glycine, arginine, and lysine?

- ☐ G, R, L                      ☐ G, A, K                      ☐ G, R, K                      ☐ Y, A, N

6- What is the heavier amino acid?

- ☐ Glycine                      ☐ Leucine                      ☐ Tryptophan                      ☐ isoleucine

7- When was the Orbitrap analyzer invented?

- ☐ 1910                      ☐ 1953                      ☐ 2000                      ☐ 2015

8- What is the SI unit of mass?

- ☐ Dalton [Da]                      ☐ Unified atomic mass [u]                      ☐ Thomson [Th]                      ☐ Dimensionless

9- What is the nominal mass of insulin ( $C_{257}H_{383}N_{65}O_{77}S_6$ )?

- ☐ 5801 u                      ☐ 5797 u                      ☐ 5803.638 u                      ☐ 5899.621 u

10- What is the charge state of a positive ion with its isotopic peaks separated by 0.25?

- ☐  $z = 1$                       ☐  $z = 2$                       ☐  $z = 3$                       ☐  $z = 4$

11- A mass spectrometer achieves an accuracy of 5 ppm. What is the  $\Delta m$  in Da at  $m/z = 1000$ ?

- ☐ 0.001 Da                      ☐ 0.002 Da                      ☐ 0.005 Da                      ☐ 0.01 Da

12- What is ESI?

- ☐ An analyzer      ☐ A detector      ☐ An ionization method      ☐ None of those

13- The ESI spectrum of a protein gives a charge state envelope. What is charge state of my peak  $m_1$  at  $m/z = 1199.08$  knowing that the next peak of lower charge is  $m_2$  at  $m/z = 1332.20$ ?

- ☐  $z_1 = 8$       ☐  $z_1 = 9$       ☐  $z_1 = 10$       ☐  $z_1 = 8$

14- From question #13, what is the average molecular mass of this protein?

- ☐ 11980.81 Da      ☐ 12580.76 Da      ☐ 14960.23 Da      ☐ 10279.67 Da

15- What flow rate is typical of the micro-spray mode?

- ☐ 10  $\mu\text{L}/\text{min}$       ☐ 300 nL/min      ☐ 1  $\mu\text{L}/\text{min}$       ☐ 1 nL/min

16- What does MALDI stand for?

- ☐ Matrix-assisted laser desorption/ionization      ☐ Metal-assisted laser desorption/ionization      ☐ Metal-assisted light desorption/ionization      ☐ Matrix-assisted light dissolution/ionization

17- What mass analyzer can offer the best resolution?

- ☐ Quadrupole      ☐ TOF      ☐ FTICR      ☐ Orbitrap

18- In an Orbitrap, the ions are trapped in an ...?

- ☐ Magnetic field      ☐ Electrostatic field      ☐ Ions are not trapped      ☐ Ions fly

19- In a TOF analyzer,  $m/z$  is proportional to?

- ☐  $t$       ☐  $1/t$       ☐  $d$       ☐ None of these

20- In what type(s) of mass spectrometer can be performed MS/MS in time?

- ☐ Q-TOF      ☐ QqQ      ☐ Ion trap      ☐ FTICR

21. What type(s) of ions are generated with collision-induced dissociation?

- ☐ b      ☐ a      ☐ z      ☐ y